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BILIWG Meeting: DOE Hydrogen Quality Working Group Update and Recent Progress

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Laurel, MD

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DOE Hydrogen Quality Working Group (H2QWG) Objectives

- Develop a process to determine hydrogen quality requirements for fuel cell vehicles based on life-cycle costs
 - evaluate impact of fuel quality requirements on hydrogen production and purification costs
 - evaluate impact of contaminants on fuel cell performance, durability, and related life-cycle costs
- Identify information gaps and the R&D needed to fill those gaps
 - recommend approaches to funding and conducting the needed R&D

H2QWG has prepared a draft Roadmap and submitted it to DOE for review and comment

The H2QWG focus is on the near- to mid-term (to 2015)

- Production: distributed (forecourt) production
 - reforming of natural gas (autothermal & steam reforming)
 - reforming of renewable fuels, e.g., ethanol (i.e., E-95 & E-85)
 - electrolysis (alkaline and PEM electrolyzers)
- Purification:
 - pressure-swing adsorption, PSA (may be aided by TSA)
 - hydrogen-permeable membrane separators
- Use in fuel cell systems (no storage effects):
 - performance / cost / durability impact of
 - electrochemically active contaminants
 - inert contaminants
- Analysis and quality verification
 - available analytical technologies (mostly research laboratory)
 - standardized (commercially accepted) technologies

Draft Roadmap

Summary findings (preliminary)

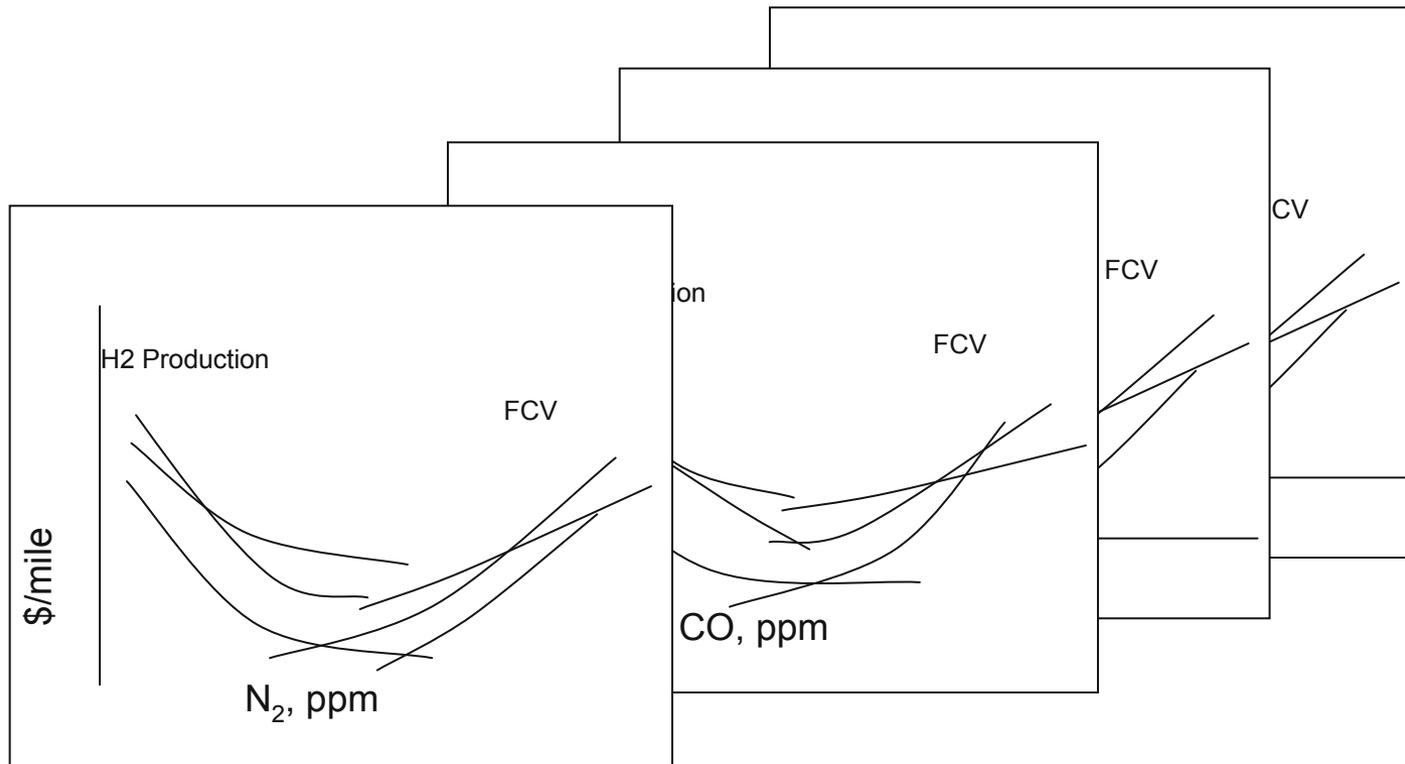
- PSA can achieve most of the H₂ impurity limits proposed by SAE / ISO, but it may add 5-20% to the cost of H₂
- PSA is ineffective for removing helium
- There are some contaminants for which PSA's effectiveness has not been reported (e.g., formic acid)
- The proposed levels for CO₂, O₂, and inert gases may be overly restrictive (based only on their effects on fuel cell performance)
- Testing and analysis may be a very significant cost factor, both for certification and for control of hydrogen quality

Draft Roadmap Recommendations (preliminary)

- Quantify the cost and performance of PSA vs. H₂ quality to determine life-cycle costs
- Quantify the effects of specific contaminants on cost and performance of fuel cells, and the costs of overcoming performance degradation
- Develop low-cost methods for gas sampling and analysis for certification and on-line quality control (and fuel quality regulation enforcement)

Modeling and experimental data are being used to assess the impact of specific fuel impurities on life-cycle costs

- Study individual contaminants
- Evaluate potentially different effects for different production / purification and fuel cell operating conditions



The H2A model is the basis for the cost of hydrogen

- Based on options / assumptions available in H2A
- The current H2A model does not reflect sensitivity to hydrogen quality
 - Add effects of hydrogen recovery and process efficiencies
- Component models are being developed to support the H2A
 - Argonne is modeling a steam reformer + PSA process
 - Results will be incorporated into H2A
 - *Look-up tables*
 - *Interface with component module*
- End Result
 - Cost of hydrogen (trend) = f (Process pathway, conditions, efficiency, contaminant level, etc.)

Typical reformate compositions from natural gas and ethanol are similar

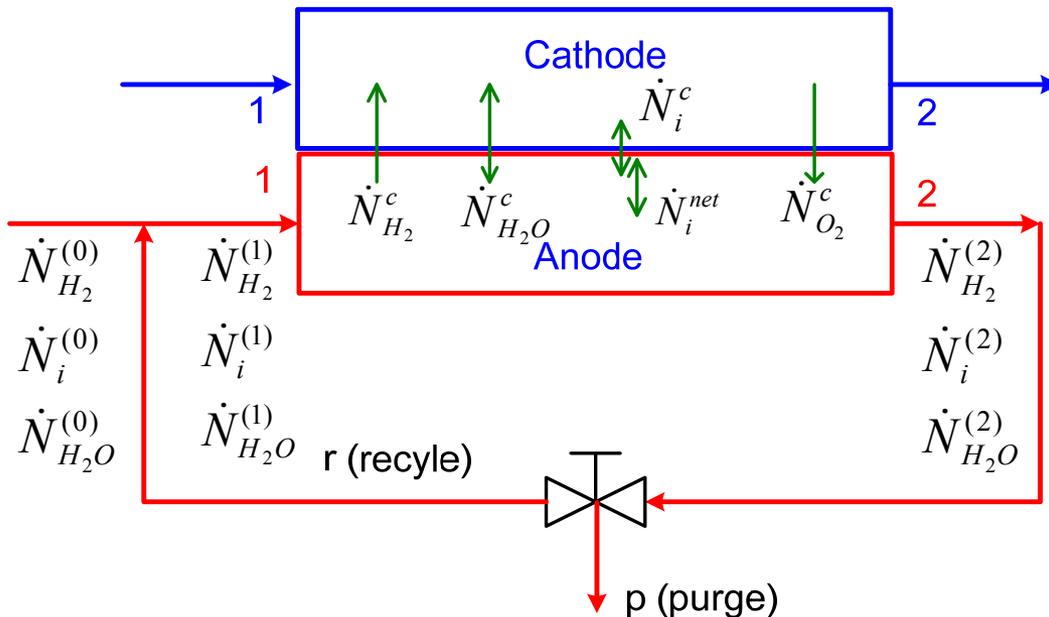
Product gas composition from steam reforming

Species	Natural Gas	Ethanol
H ₂	75–80%	70–75%
CO ₂	15–18%	22–25%
CO	0.1–4%	0.1–4%
CH ₄	0.5–3%	0.5–3%
Non-CH ₄ HCs	0.5%	0.5%
He	≤500 ppm	trace
N ₂	100 ppm	100 ppm
Ar	5 ppm	5 ppm
H ₂ S	5 ppm	1 ppm
O ₂	trace	trace
NH ₃	trace	trace

Reformate from ethanol may also have traces of organic acids and aldehydes

Fuel cell stack model: Effects of fuel impurities on fuel cell performance

- Current generation and transport in catalyst layers
- Transport of ions across PFSA membrane
- Kinetics of HOR and ORR over Pt catalyst
- Multi-species diffusion in porous media
- Water transport across PFSA membranes
- Capillary transport of water across GDL and catalyst
- 2-phase flow in gas channels



- Once-through cathode stream
- Anode gas recirculation

$$R = \dot{N}_r / \dot{N}_p$$

- Crossover of H_2 , O_2 , N_2 , H_2O and NH_3

Mechanistic models of electrocatalyst poisoning

■ Hydrogen Oxidation Reaction

- $\text{H}_2 + 2\text{M} \rightleftharpoons 2\text{M-H}$ (Dissociative Adsorption)
- $\text{M-H} \rightarrow \text{M} + \text{H}^+ + \text{e}^-$ (Electrochemical Oxidation)

■ CO Poisoning of Pt

- $\text{CO} + 2\text{M} \rightleftharpoons \text{M}_2\text{-CO}$ (Associative Adsorption on Bridge Sites)
- $\text{CO}_2 + 2\text{M-H} \rightarrow \text{M}_2\text{-CO} + \text{H}_2\text{O}$ (Reverse Water-Gas Shift)
- $\text{M}_2\text{-CO} + \text{H}_2\text{O} \rightarrow 2\text{M} + \text{CO}_2 + 2\text{H}^+ + 2\text{e}^-$ (Electrochemical Oxidation)

■ Reactions with Oxygen

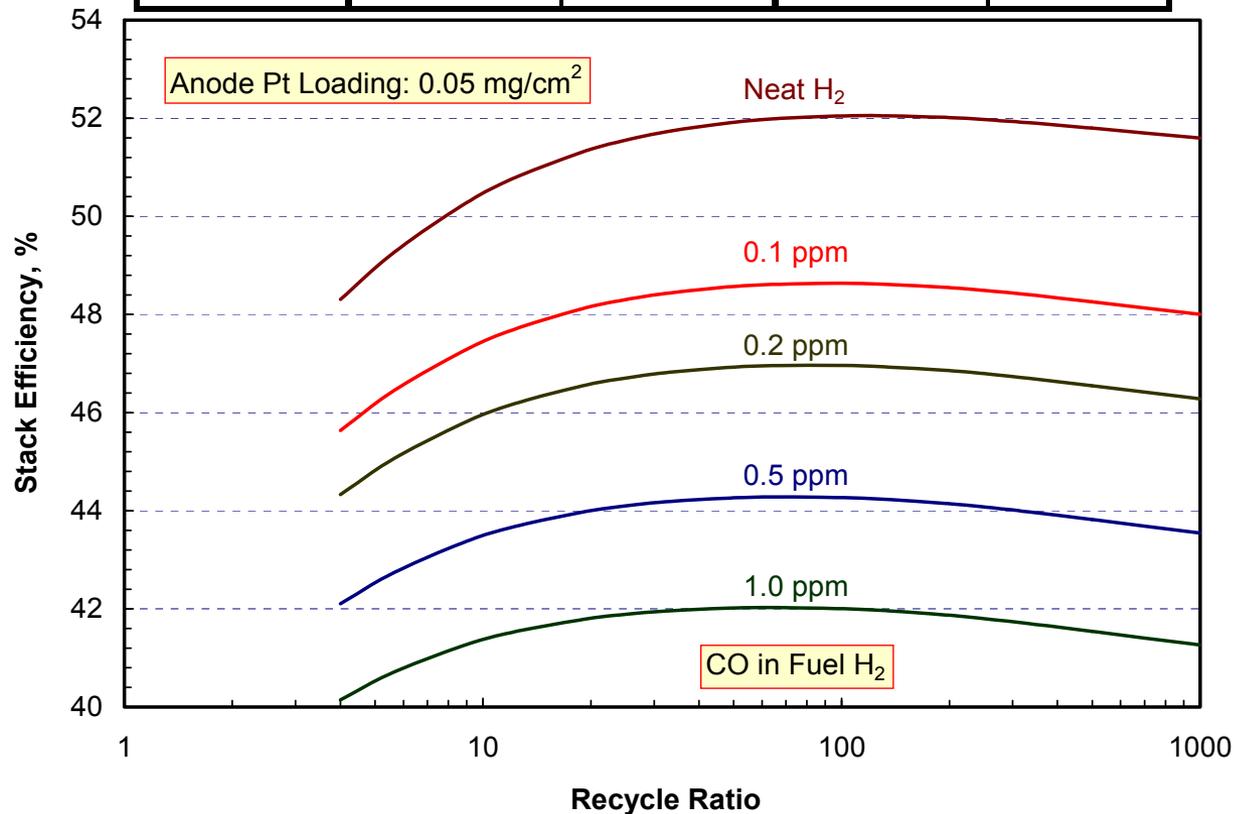
- $\text{M}_2\text{-CO} + \frac{1}{2} \text{O}_2 \rightarrow 2\text{M} + \text{CO}_2$ (CO Oxidation)
- $2\text{M-H} + \frac{1}{2} \text{O}_2 \rightarrow 2\text{M} + \text{H}_2\text{O}$ (H₂ Oxidation)

■ H₂S Poisoning of Pt

- $\text{M} + \text{H}_2\text{S} \rightleftharpoons \text{M-H}_2\text{S}$ (Reversible Associative Adsorption)
- $\text{M-H}_2\text{S} + \text{M-H} \rightarrow \text{M}_2\text{S} + 3/2\text{H}_2$ (Irreversible Dissociation)
- $\text{M}_2\text{S} + 2\text{H}_2\text{O} \rightarrow 2\text{M} + \text{SO}_2 + 4\text{H}^+ + 4\text{e}^-$ (Electrochemical Oxidation)

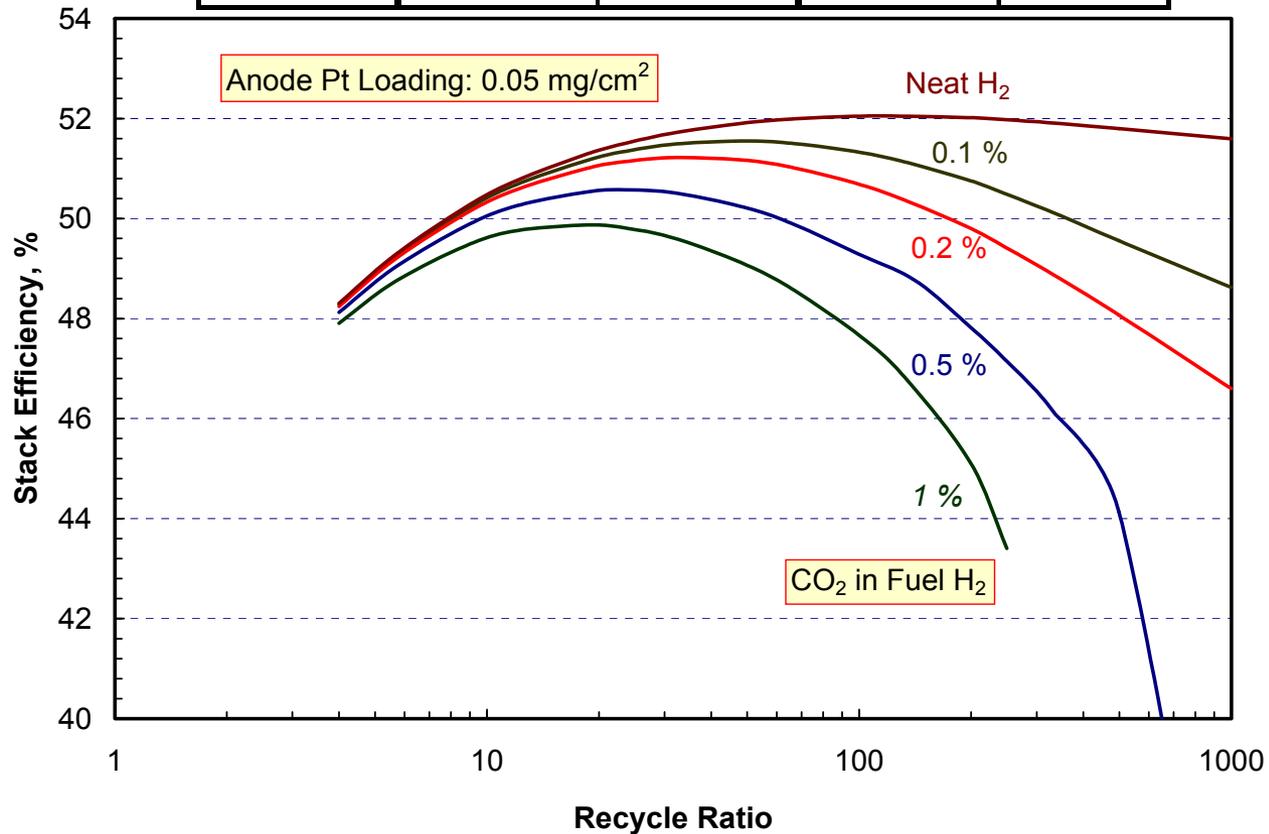
CO poisoning: Effect of Pt loading on fuel cell performance degradation

Pt Loading	0.4 mg/cm ²		0.05 mg/cm ²	
CO, ppm	ΔV	$\Delta \eta$	ΔV	$\Delta \eta$
0.1	18.4	1.5	42.3	3.4
0.2	28.1	2.2	63.3	5.1
0.5	44.1	3.6	95.9	7.8
1.0	60.5	4.9	124.3	10.0



CO₂ poisoning: Effect of Pt loading on fuel cell performance degradation

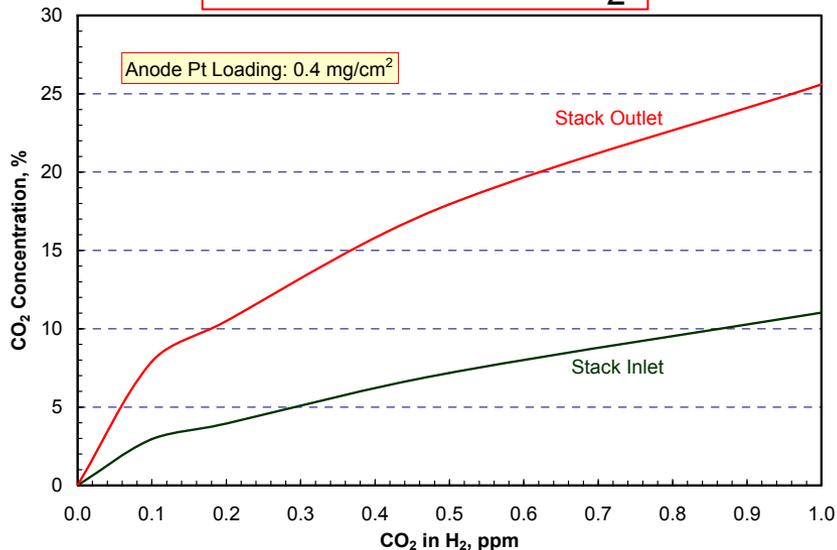
Pt Loading	0.4 mg/cm ²		0.05 mg/cm ²	
CO ₂ , %	ΔV	$\Delta\eta$	ΔV	$\Delta\eta$
0.1	5.7	0.7	4.2	0.4
0.2	7.8	1.1	5.6	0.8
0.5	14.3	1.8	11.1	1.4
1.0	20.5	2.5	17.4	2.1



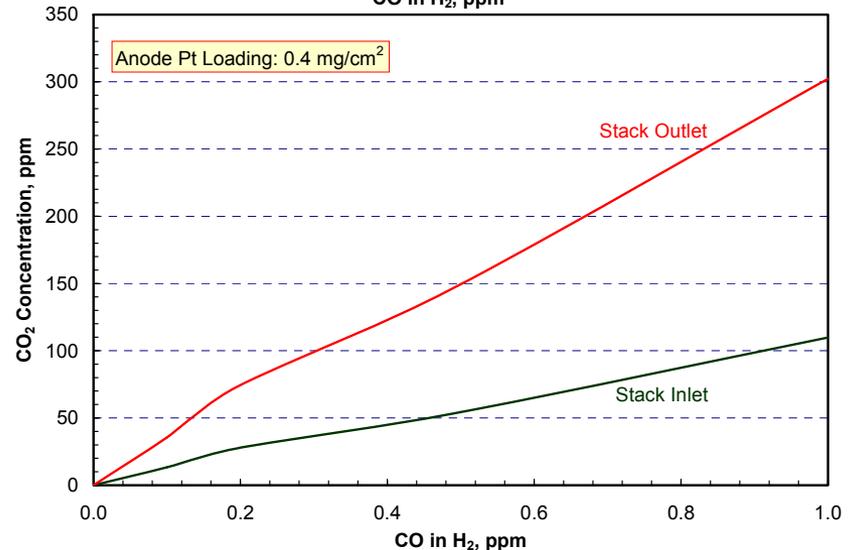
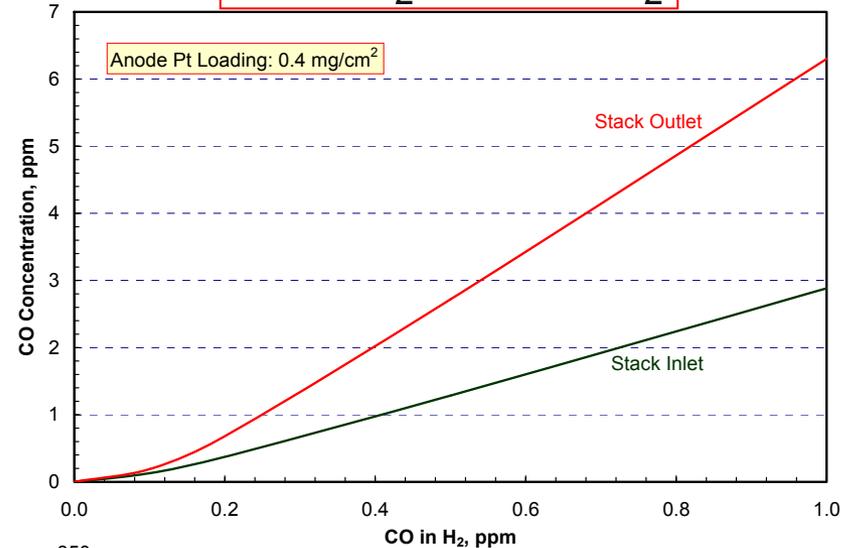
Buildup of CO and CO₂ in the recirculating anode gas

- Fraction of CO at stack inlet that is converted to CO₂ by O₂ crossing over from cathode (internal air bleed, 50- μ m membrane)
 - <50% for 0.1-ppm CO in fuel H₂
 - ~20% for 1-ppm CO in fuel H₂

No CO in Fuel H₂

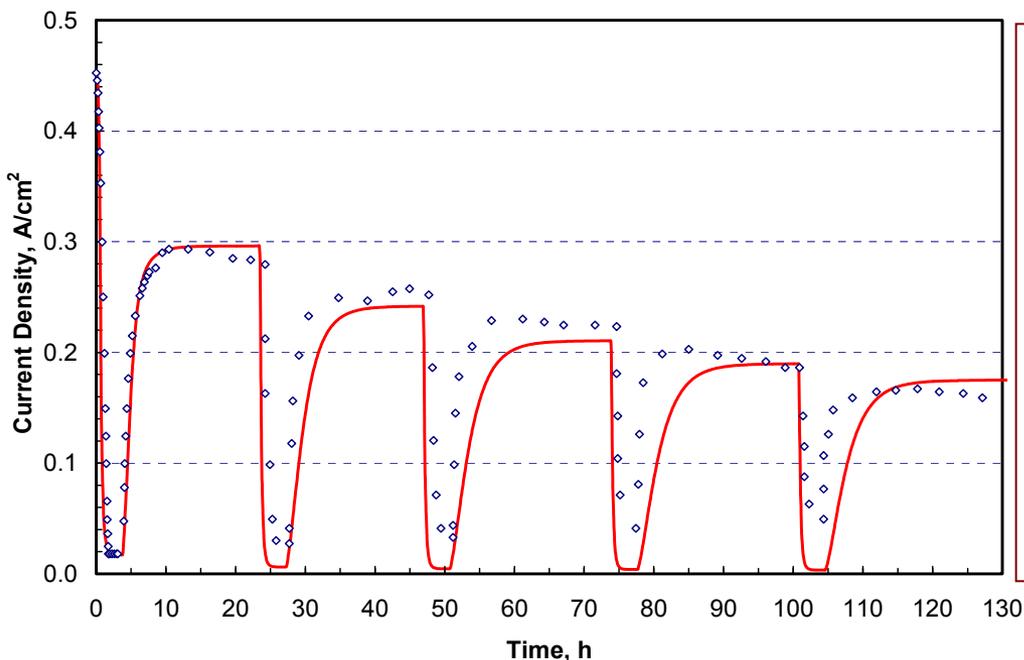


No CO₂ in Fuel H₂



H₂S poisoning: Transient poisoning and recovery

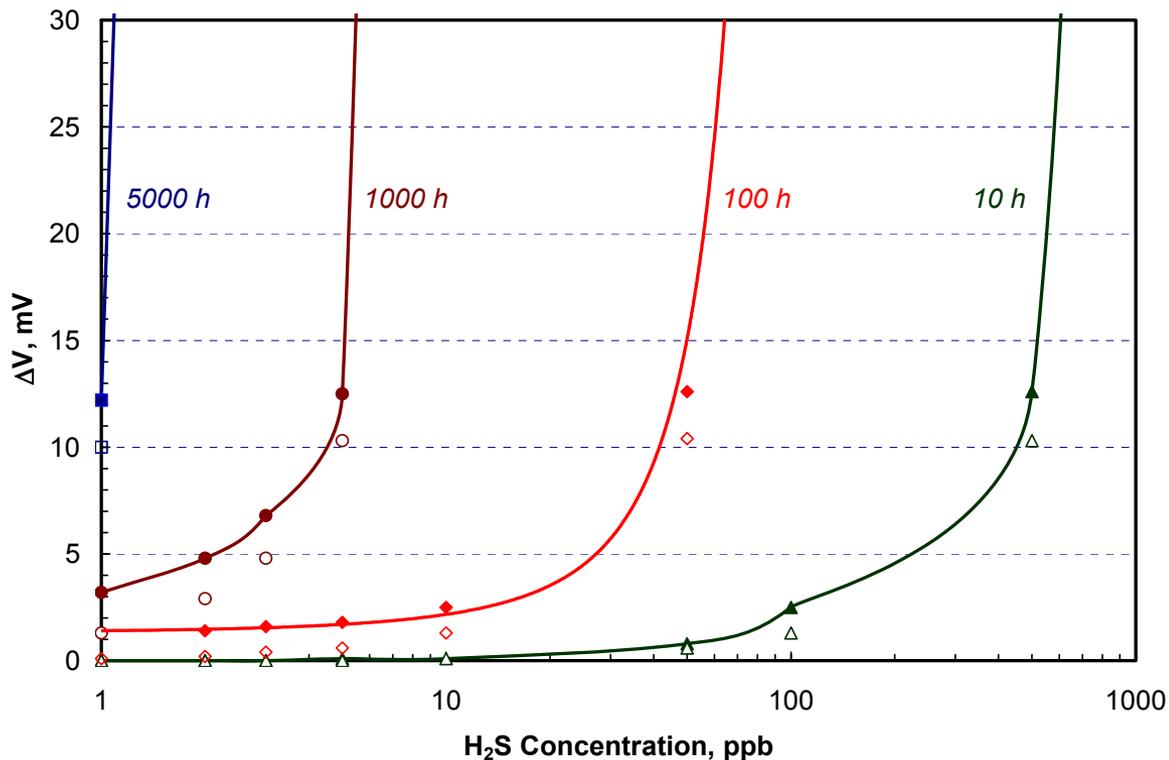
- Data from Mohtadi, PhD thesis, USC (2004)
- Kinetic constants for H₂S reactions derived from measured transient poisoning and recovery response of current density
- Partial recovery in neat H₂, progressive degradation over five cycles



- Gore PRIMEA MEA Series 5510
- 25 μm membrane
- 0.4 mg/cm² Pt on anode
- 0.4 mg/cm² Pt on cathode
- Poisoned by 50-ppm H₂S for 3.8 h
- Recovery in neat H₂ for 24 h
- Constant cell voltage: 0.69 V
- 70°C, 101 kPa

Effect of H₂S dosage on cell voltage decline

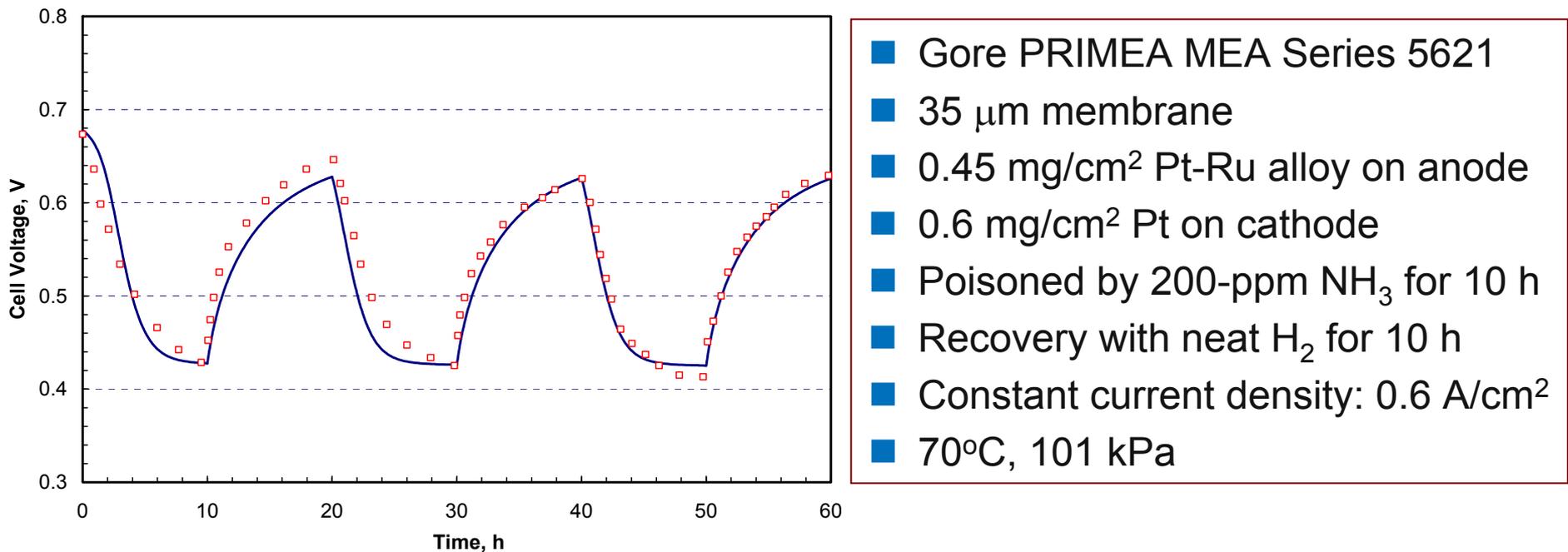
- H₂S concentration needs to be <1 ppb to limit decrease in cell voltage at 1.05 A/cm² to 10 mV after 5000 h



- 1.05 A/cm² current density
- 70% H₂ utilization per pass
- 50% O₂ utilization
- Open Symbols (R=10)
- Closed Symbols (R=100)

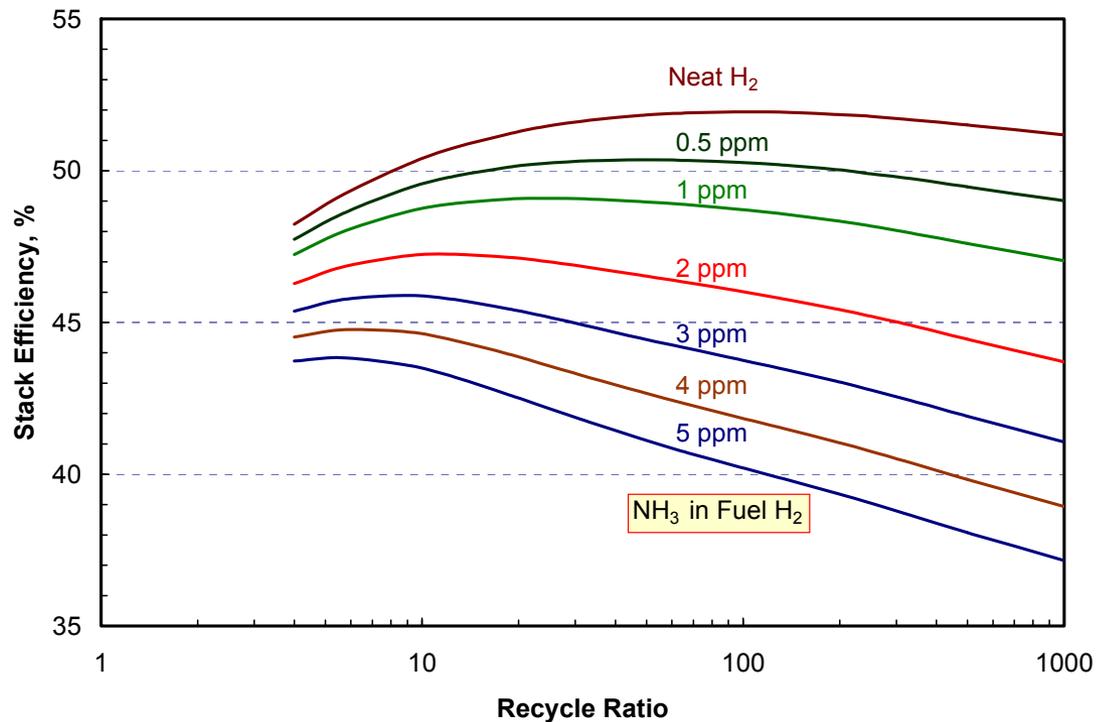
NH₃ poisoning: Transient poisoning and recovery

- Data from Soto et. al., Electrochemical and Solid-State Letters, 6 (7) A133-A135, 2003
- Kinetic constants for NH₃ uptake reactions derived from measured transient poisoning and recovery response of cell voltage



Effect of NH_3 buildup on stack efficiency

NH_3 in Fuel H_2	Recycle Ratio	Purge	$\Delta\eta$
0.5 ppm	50	~2%	1.6%
1 ppm	25	~4%	2.9%
2 ppm	12	~7%	4.7%
5 ppm	6	~15%	8.1%



■ Results at constant 1.05 A/cm^2

Held a hydrogen quality modeling workshop at Argonne on August 30-31, 2007

- Purpose: to describe models being developed to assess impurity effects on fuel cell performance and durability (and hydrogen purification by PSA)
 - Describe the significant components and processes in the models
 - Provide details of input parameters
 - *Sensitivity of output results to input parameters*
 - Define data needed to validate / refine models
 - Develop mutual understanding between modelers and experimentalists on what is achievable
 - Identify limitations / capabilities of modeling and experimentation
 - *How to reduce limitations, increase capabilities*
 - Develop specific means for maintaining continuing interactions among this research community

Status of impurities effects modeling and future work

- We have developed a framework for modeling the effects of impurities (site blockage, ion exchange, HOR and ORR reaction kinetics for hydrogen oxidation and oxygen reduction, etc.)
 - Quantitatively predict effects of impurities, not just to explain experimental observations
 - Assess effects of simultaneous presence of multiple impurities
 - Simulate steady-state and dynamic effects
 - Modify / update mechanisms as additional data become available
 - Extend models to other fuel impurities
 - Extend models to air impurities

Summary

- The H2QWG has prepared and submitted to DOE a draft roadmap for continuing activities to address issues of hydrogen quality for automotive fuel cell systems
- DOE contaminant effects projects are being complemented by the fuel cell system modeling work at Argonne
- There is close interaction between these activities and the related work of SAE and ISO by participation in Working Group meetings and workshops