



... for a brighter future

Universal Membrane Classification Scheme

How can DOE maximize the return on
High Temperature PEM Membrane Research
Efforts?

High Temperature Membrane Working Group

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U.S. Department
of Energy

UChicago ►
Argonne_{LLC}

A U.S. Department of Energy laboratory
managed by UChicago Argonne, LLC

Huge variety of possible polymer membranes to investigate

- Hydrocarbon, Fluorocarbon, Inorganic
- Branched, Comb, hyperbranched, dendritic,,
- With additives
- With structural supports,

What avenues should be pursued?

Which are dead ends?

How do we decide?



Need a way to organize/manage information gathered

- We are developing a membrane classification scheme to help
 - Track different types of approaches
 - Determine which strategies are most fruitful
 - Determine which strategies are dead ends
 - Improve our understanding of proton transport
 - Help maximize our return on HTM research

Contributors

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Categorized based on Strategies for conduction and membrane morphology

- 1) Type of conduction mechanism
 - Aqueous
 - Non-aqueous

- 1) Plan to control the membrane structure/morphology
 - Molecular
 - Additive
 - Micro/nanoengineering

Morphology approaches

- Molecular approach- use molecular design to provide phase separation to improve mechanical properties and proton conduction under low RH conditions by concentrating acid groups in hydrophilic phase
- Additive approach- use of additives to maintain some water at additive surfaces at high temperatures, allowing for conduction at high temperature-low RH,
- micro/nano-engineering approach - by designing physical structure of materials (ie pores, or fibers) engineer material to provide domains of high-conductivity, which will operate at high-temp low RH (and be freed of requirements for mechanical integrity/strength) and separate domains of high-strength.

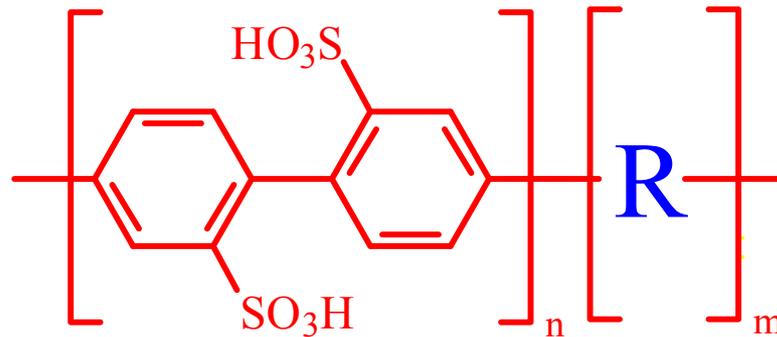
Level 1 Categorization

Conduction Mechanism	Morphology	Molecular approach	Additive approach	micro/nano-engineering approach
Aqueous Conduction-				
Non aqueous conduction				

Molecular approach example

Poly (p-Phenylene Sulfonic acids)

Case Western Reserve University Approach: Use Rigid Rod, Liquid Crystalline polymers with the acid groups directly attached to the backbone. Use poly(p-phenylene sulfonic acid); it can't hydrolyze. Use bulky comonomers to generate free volume.



R = bulky comonomer

Additive approach example

■ **Florida Solar Energy Center Approach**

SPEEK and SPEKK-based PTA Composite Membranes

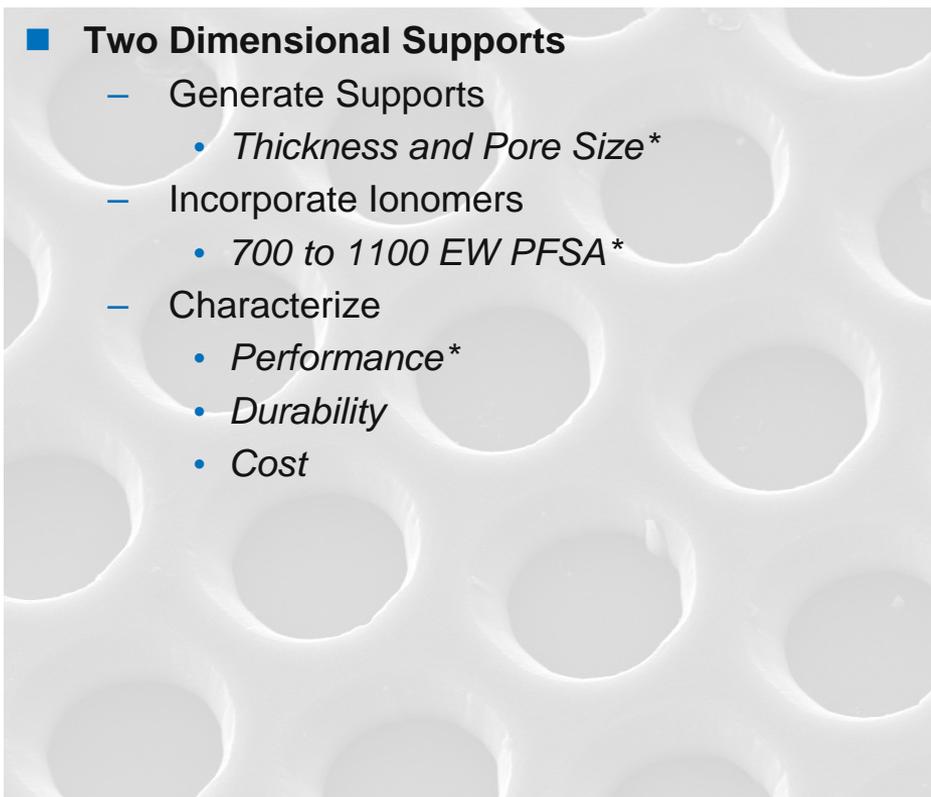
Poly[perfluoro-sulfonic acid] - Teflon® -phosphotungstic acid (PTA) membranes

Micro/nanoengineering approach example

■ Two Dimensional Supports

- Generate Supports
 - *Thickness and Pore Size**
- Incorporate Ionomers
 - *700 to 1100 EW PFSA**
- Characterize
 - *Performance**
 - *Durability*
 - *Cost*

Gener Electrochemical Systems Approach:
Lower EW of perfluorosulfonic acid ionomers to increase low RH conductivity and support the ionomer with two and three-dimensional non-ionic materials



Mag:700 kV:20 plasma clean, bottom surface 10 μm

Level 2

Conduction Mechanism		Morphology	Molecular approach		Additive approach		micro/nanoengineering approach	
Aqueous Conduction-			Structured polymers	Random polymers	Organic additive	Inorganic additive	Structured support	Structured ionomer
	Sulfonic acids							
	hydrous metal oxides							
	Sulfonyl imides							
Non aqueous conduction								
	Inorganic oxides							
	P-acids							
	Heterocyclic bases							
	Ionic liquids							

Categorization- Level 3

Conduction Mechanism	Morphology	Molecular approach					Additive approach					micro/nano-engineering approach				
		Random Polymer		Structured Polymer			Organic additive	Inorganic additive				Structured support		Structured ionomer		
		Mono Polymer	Polymer blends	Graft	Block	Self assembling domains		SiO2	TiO2	ZrHPO4	HPA	Zeolite	Porous	Fibrous	Porous	Fibrous
Aqueous Conduction-																
	<i>Sulfonic acids</i>															
		Fluorosulfonic acids														
		HC-Sulfonic acids														
	<i>Hydrous Metal oxides</i>															
		SiO2														
		TiO2														
	Sulfonyl imides															
		Fluorosulfonyl imides														
Non aqueous conduction																
	<i>Inorganic oxides</i>															
		HPAs														
		Metal Oxides														
		ZrHPO4														
	<i>P-acids</i>															
		Phosphoric acids														
		Phosphonic acids														
	<i>heterocyclic bases</i>															
		imidazoles														
		triazoles														
		oxazoles														
	<i>ionic liquids</i>															
		imidazolium salts														
		quaternary ammonium salts														

Level 4

Aqueous Conduction-			
	<i>Sulfonic acids</i>		
		Fluorosulfonic acids	
			aromatic backbone
			non-aromatic backbone
		HC-Sulfonic acids	
			aromatic backbone
			non-aromatic backbone

Level 5

Aqueous Conduction-			
	<i>Sulfonic acids</i>		
		Fluorosulfonic acids	
			aromatic backbone
			Sulfonated ring
			linear sidechain
			branched sidechain
			non-aromatic backbone
			linear sidechain
			branched sidechain
		HC-Sulfonic acids	
			aromatic backbone
			Sulfonated ring
			linear sidechain
			branched sidechain
			non-aromatic backbone
			linear sidechain
			branched sidechain

Categorization- Level 3

Morphology		Molecular approach					Additive approach					micro/nano-engineering approach			
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	imidazolium salts														
	quaternary ammonium salts														

We need your help!

- Identify where your projects belong in the scheme
 - Share as much information as possible!
- Identify testing concerns

Why should you help?

- Input in to possible testing requirements
- Eliminate need to develop individual tests for each membrane
- Identify subgroups for collaboration

Needs/concerns may be different for different categories.

- Micro/nano-engineering approach
 - can be highly nonisotropic
 - need thru plane measurements of conductivity?

- Additive approach
 - migration of additives could be problematic
 - concerns about additive leaching
 - additives may affect peroxy radical formation, decomposition

Effects of Strategy on membrane properties

- How does the membrane strategy affect conductivity?
- Transport properties – water, O₂, H₂ ?
- Chemical durability?
- Mechanical durability?
- Integration with catalyst and catalyst layer?

How does strategy affect testing?

Do we need new/different tests for some of the strategies?

Why should DOE continue to look at multiple approaches?

- Different risks/benefits for the different conduction mechanisms
 - Aqueous mechanisms-less risky for improving conductivity at moderately lower RH and moderate temperatures, more risky at very low RH and higher temperatures. Likely to provide some intermediate benefits
 - Nonaqueous mechanisms – Potential for water-free conduction so conductivity at extremely low RH possible, higher risk



New Targets Proposed for Membranes

Characteristic	Units	2006 status	2005 target	2010 target	2015 target
Maximum operating temperature	°C	80	120	120	120
Area specific resistance at: Maximum operating temp and water partial pressures from 40 – 80 kPa	Ohm cm ²	0.03	0.02	0.02	0.02
80°C and water partial pressures from 25 - 45 kPa	Ohm cm ²	0.03	0.02	0.02	0.02
30°C and water partial pressures up to 4 kPa	Ohm cm ²	0.04	0.03	0.03	0.03
-20°C and water partial pressures up to 0.1 kPa	Ohm cm ²	0.3	0.2	0.2	0.2
Oxygen cross-over ^a	mA/cm ²	5	5	2	2
Hydrogen cross-over ^a	mA/cm ²	5	5	2	2
Cost ^b	\$/m ²	15 ^c	200	20	20
Durability with cycling At operating temp ≤80°C	hours	~2000 ^d	2000	5000 ^e	5000 ^e
At operating temp >80°C	hours	(not avail. ^f)		2000	5000 ^e
Unassisted start from	°C	-20	-30	-40	-40
Thermal cyclability in presence of condensed water		Yes	Yes	Yes	Yes

What we want a membrane to do



Thank you for your attention!

